

VEER SURENDRA SAI UNIVERSITY OF TECHNOLOGY, BURLASemester: **Third****LESSON PLAN**Subject: **Chemistry III (Basic Physical-I)**
(CH-211)Session: **Odd 2016–2017****Theory/Sessional**Branch/Course: **Int.M.Sc.(5yr)**Name of the Faculty Member: **Dr. Monalisa Mohapatra**

Period	Module/ Number	Topics to be Covered	Remarks/Sign. of Faculty Member
1	I	Dilute Solutions: Vapour pressure	
2	I	Thermodynamic derivation of vapour pressure	
3	I	Colligative properties	
4	I	Raoult's law	
5	I	Thermodynamic derivation of laws relating to elevation of boiling point	
6	I	Depression of freezing point	
7	I	Osmotic pressure	
8	I	Ideal and nonideal solution	
9	I	Association and dissociation	
10	II	Homogenous equilibrium: Law of mass action	
11	II	Thermodynamic derivation of the expression for equilibrium constant	
12	II	Different forms of equilibrium constants	
13	II	Lechatelier's principle, Illustrations with some gaseous reactions	
14	II	Effect of temperature on equilibrium Vant-hoff's equation and its integration	
15	III	Chemical kinetics: Order and molecularity	
16	III	Kinetics of 1st order reactions	
17	III	Kinetics of 2nd order reactions	
18	III	Simple opposing (A&B) reaction	
19	III	Consecutive or Sequential (A-B-C) reaction	
20	III	Chain reaction ($H_2 + Br_2$)	

21	III	Effect of temperature on reaction rate	
22	III	Collision theory of reaction rate	
23	III	Qualitative treatment of transition theory	
24	IV	Thermodynamic concept: Heat content and heat capacity	
25	IV	Work done for Isothermal change	
26	IV	Work done for Adiabatic change	
27	IV	Work done for Ideal gases	
28	IV	Work done for van der Waal gases	
29	IV	Thermochemistry: Heat changes in chemical reactions	
30	IV	Hess's law	
31	IV	Kirchoff's equation	
32	IV	2nd law of thermodynamics: Spontaneous process	
33	IV	Carnot's theorem and Carnot's cycle	
34	IV	Efficiency of heat engine	
35	IV	Entropy changes in reversible processes	
36	IV	Entropy changes in irreversible processes	
37	IV	Free energy Condition for equilibrium	
38	IV	Work function Condition for equilibrium	
39	IV	Claypeyron and Clausius equation	
40	IV	Gibbs Helmholtz equation	

Signature of Faculty Member:

Date:

Counter Signature of H.O.D.

VEER SURENDRA SAI UNIVERSITY OF TECHNOLOGY, BURLA

LESSON PLAN

Semester: **3rd**

Subject: **ENVIRONMENTAL SCIENCE (CH-212)**

Session: **Odd 2016–2017**

Theory/Sessional

Branch/Course: **Int. M. Sc.**

Name of the Faculty Member: **Prof. R. B. Panda**

Period	Module/ Number	Topics to be Covered	Remarks/Sign. of Faculty Member
1-2	I	Scope of Ecology, Component of Ecosystem	
3-4	I	Concept and Definition of Environmental Pollution	
5-6	I	Types and Classification of Pollution, Pollution and Source of Pollution. History of Major Pollution Episodes	
7-8	I	Concept of Hydrosphere, Concept of Atmosphere and its Pollution	
9-10	I	Major and Minor Pollutants in Atmosphere (SO _x , NO _x , CO ₂ , Fluoride, Hydrocarbon)	
11-12	II	Classification and Types of Water Pollution	
13-14	II	Industrial Waste, Municipal Waste	
15-16	II	Agriculture Chemicals, Oil Pollution	
17-18	II	Heavy Metals Ground Water Pollution (Mercury).	
19-20	II	Heavy Metals Ground Water Pollution (Lead, Arsenic).	
21-22	III	Acid Rain, Photochemical Smog, Greenhouse Effect	
23-24	III	Ozone Layer Depletion, Eutrophication, Ecological Magnification	
25-26	III	Concept of Lithosphere, Sources of Soil Pollution	
27-28	III	Pollution Effect of Pesticides and Fertilizer in Soil	
29-30	III	Use of Flora for Control of Pollution	
31-32	IV	Pollution by radiation: Sources of Radioactive Pollution	
33-34	IV	Effect of Radiation Protection and Control from Radiation	
35-36	IV	Disposal of Radioactive Waste	
37-38	IV	Sources of Noise, Noise Levels in Decibel Scale	
39-40	IV	Effect of Noise on Human Health, Prevention and Control of Noise	

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VEER SURENDRA SAI UNIVERSITY OF TECHNOLOGY, BURLA

LESSON PLAN

Semester: **4th**

Session: Even **2016–2017**

Branch/Course: **Int. M. Sc.**

Subject: **Basic Organic I (CH-221)**

Theory/Sessional

Name of the Faculty Member: **Prof. R. B. Panda**

Period	Module /Number	Topics to be Covered	Remarks/Sign. of Faculty Member
1-2	I	Formation, Stability and Structure of Free Radicals and Carbenes	
3-4	I	Nitrene, Enamine, and Benzyne	
5-6	I	Reaction in Organic Compounds, <i>Classification of Reactions:</i> Substitution, addition	
7-8	I	Elimination, Electron Transfer Reaction, Molecularity, Order of Reactions	
9-10	I	Transition State and Intermediates, Nucleophiles and Electrophiles	
11-12	II	<i>Configurational Isomerism:</i> Optical Isomerism: Introduction, Conditions for Optical Activity, Optical Rotation, Specific Rotation	
13-14	II	D & L Convention, R & S Notations, Optical Isomers of Lactic, and Tartaric Acids	
15-16	II	Enantiomers and Diastereomers, Threo and Erythro Nomenclature, Meso Compounds, Racemic Modification, Methods of Resolution	
17-18	II	Introduction, Structural Requirement (Cis & Trans, Syn & Anti), E- Z Convention, Configuration of Oximes.	
19-20	II	Introduction, Conformations of Ethane and n-Butanes & Cyclohexane, Baeyer Strain Theory	
21-22	III	<i>Grignard's Reagent:</i> Preparation, Structure, Synthetic Uses	
23-24	III	<i>Esters Containing Active Methylene Groups:</i> Acetoacetic Ester: Synthesis, Synthetic Uses, structure and Keto-Enol Tautomerism	
25-26	III	<i>Malonic Ester:</i> Preparation and Synthesis Uses.	
27-28	III	Classification, Configuration of Sugars, Glucose and Fructose (Occurrence, Reaction: Osazone formation with Fehlings Solution	
29-30	III	Mutarotaion, Elucidation of Structure of D-Glucose (Open Chain and Ring Structure)	
31-32	IV	Five Membered heterocyclics: Pyrrole, Thiophene and Furan: Synthesis (from Sugar, Dicarbonyl Compound)	
33-34	IV	operties (Aromaticity, Electrophilic Substitution Reactions) of Five Membered heterocyclics	
35-36	IV	<i>Acyclic Compounds:</i> Preparation, Reactions and Stability.	
37-38	IV	<i>Aryl Nitrogen Compounds</i> Nitro Hydrocarbons, Preparations	
39-40	IV	<i>Aryl Nitrogen Compounds:</i> Properties, Reduction of Nitro Benzene, TNT, Amines	

Signature of Faculty Member:

Date:

Counter Signature of H.O.D.

VEER SURENDRA SAI UNIVERSITY OF TECHNOLOGY, BURLASemester: 5th**LESSON PLAN**Subject: **Basic Organic-II
(CH-311)**Session: **Odd****Theory/Sessional**Branch/Course: **Int.M.Sc.(5yr)**Name of the Faculty Member: **Dr A. K. Panda**

Period	Module/ Number	Topics to be Covered	Remarks/Sign. of Faculty Member
1	I	Types of mechanisms, types of reactions	
2	I	Aliphatic substitution reaction: SN1 Kinetics, stereochemistry, structural and environmental aspects), Neighboring group participation reactions	
3	I	Aliphatic substitution reaction: SN2 Kinetics, stereochemistry, structural and environmental aspects)	
4	I	Aliphatic substitution reaction: SNi reactions, Neighboring group participation reactions	
5	I	Conformational Isomerism: Introduction conformations of cyclohexane	
6	I	Conformational analysis, mono and di substituted cyclohexanes	
7	I	Dyes: Colour and constitution	
8	I	Classification of dyes	
9	I	Chemistry and synthesis of methyl orange	
10	II	Chemistry and synthesis of Bismarck brown.	
11	II	Heterocyclic compounds: Six membered heterocycles	
12	II	Pyridine (Preparation and Reactions)	
13	II	Quinoline (Preparation and Reactions)	
14	II	Fused heterocycles: Urides	
15	II	Elucidation of structure of Uric acid	
16	II	Purines	
17	II	Indigo: Structure and Use	
18	II	Polynuclear Hydrocarbons: Naphthalene and Anthracene).	
19	II	Addition and electrophilic substitution reactions of Naphthalene and Anthracene	
20	II	Elucidation of their structures Naphthalene and Anthracene	
21	III	Alkaloids: Introduction	

22	III	Elucidation of structure of nicotine	
23	III	Synthesis of nicotine	
24	III	Elucidation of structure of papavarine	
25	III	Synthesis of papavarine	
26	III	Terpenes: Introduction Isoprene rule	
27	III	Elucidation of the structure of the camphor	
28	III	Synthesis of Camphor	
29	III	Vitamines: Introduction	
30	III	Elucidation of the structure of the Vitamine C	
31	IV	Mechanism and applications of Molecular rearrangements: Pinacole-pinacolone	
32	IV	Mechanism and applications of Demnzenov	
33	IV	Mechanism and applications of Dienone-phenol	
34	IV	Mechanism and applications of Beckmann and Benzidene.	
35	IV	Principle mechanism and application of Name reactions: Diels-Alder, Fries	
36	IV	Principle mechanism and application of Michael, Mannich reaction	
37	IV	Principle mechanism and application of Reformatsky, Cleisen and Dickmann reactions	
38	IV	Preparation properties and application of Reagent: LiAlH_4 , NaBH_4	
39	IV	Preparation properties and application of Reagent: HIO_4 , $\text{PhI}(\text{OAc})_4$	
40	IV	Preparation properties and application of Reagent: PCC and DCC	

Signature of Faculty Member:

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VEER SURENDRA SAI UNIVERSITY OF TECHNOLOGY, BURLA

LESSON PLAN

Semester: 5th

Subject: **Solid State Chemistry (CH-312)**

Session: Odd 2016–2017

Theory/Sessional

Branch/Course: Int. M. Sc.

Name of the Faculty Member: **Dr. Biswa Nath Ghosh**

Period	Module/ Number	Topics to be Covered	Remarks/Sign. of Faculty Member
1-2	I	Description of Crystalline and Amorphous Solids; Crystal Systems	
3-4	I	Point Groups, Methods of Characterizing Crystal Structure - Powder X-ray Diffraction, Electron and Neutron Diffraction	
5-6	I	Close Packing - HCP and CCP, Packing Efficiency, Radius Ratios	
7-8	I	Description of Solids-NaCl, ZnS, Na ₂ O, CdCl ₂ , Wurtzite, Nickel Arsenide	
9-10	I	Description of Solids-CsCl, CdI ₂ , Rutile and Cs ₂ O, Perovskite ABO ₃ , K ₂ NiF ₄ , Spinel.	
11-12	II	<i>Preparative Methods:</i> Solid State Reaction, Chemical Precursor Method	
13-14	II	<i>Preparative Methods:</i> Co-precipitation, Sol-Gel, Metathesis	
15-16	II	Self-propagating High Temperature Synthesis, Ion Exchange Reactions	
17-18	II	Intercalation/Deintercalation Reactions; Template Synthesis	
19-20	II	Hydrothermal: High Pressure Synthesis.	
21-22	III	<i>Characterization:</i> Thermal Analysis - TGA, DTA, DSC	
23-24	III	Electrical Properties: Band Theory of Solids -Metals and their Properties	
25-26	III	Semiconductors - Extrinsic and Intrinsic; Hall effect	
27-28	III	Thermoelectric Effects (Thomson, Peltier and Seebeck)	
29-30	III	Insulators - Dielectric, Ferroelectric, Pyroelectric and Piezoelectric Properties; Ionic Conductors.	
31-32	IV	<i>Magnetic Properties:</i> Dia, Para, Ferro, Ferri, and Antiferro	
33-34	IV	Magnetic Types; Soft and Hard Magnetic Materials; Select Magnetic Materials such as Spinel, Garnets and Perovskites, Hexaferrites	
35-36	IV	Lanthanide-Transition Metal Compounds; Magnetoresistance.	
37-38	IV	<i>Optical Properties:</i> Luminescence of d- and f- block ions; Structural Probes;	
39-40	IV	Up and Down Conversion Materials	

Signature of Faculty Member:	
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VEER SURENDRA SAI UNIVERSITY OF TECHNOLOGY, BURLA

Semester: 5th M.Sc

Lesson Plan

Subject: **Green Chemistry (CH-313)**

Session: **Odd 2016-17**

Theory

Course: **5-Year Int.M.Sc (Chemistry)**

Name of the Faculty Member: **Dr. Ramakrishna D.S.**

Period	Module/ Number	Topics to be Covered	Remarks/Sign. of Faculty Member
1	I	Green alternatives to synthesis Organic Transformations: Principles of green chemistry	
2	I	Planning of a green organic synthesis	
3	I	Aqueous phase Transformations	
4	I	p-Acetylaminophenol	
5	I	3 Aminopyridine, Anthranilic acid	
6	I	Benzoin, n-butyl bromide	
7	I	cycloheptanone, 2,4- dihydroxy benzoic acid,	
8	I	Hippuric acid	
9	I	Pinacolone	
10	I	Miscellaneous transformations in water	
11	III	Transformations Using Phase transfer catalysis	
12	III	Benzoic acid, benzonitrile,	
13	III	n-butyl benzyl ether,	
14	III	4,6-dimethyl-3-phenyl coumarin,	
15	III	flavone, phenylisocyanide, salicylaldehyde,	
16	III	Applications: Benzoin condensation,	
17	III	Darzen's reaction, Michael reaction,	
18	III	Williamson ether synthesis	
19	III	Wittig reaction,	
20	III	Wittig Horner reaction.	

22	II	2-Allyl phenol, anthraquinone,	
23	II	benzil, 2,5-dimethylpyrrole, flavone,	
24	II	Applications: Oxidation of alcohols to carbonyl compounds,	
25	II	oxidation of sulphides to sulphoxides, Pinacol-Pinacolone rearrangement,	
26	II	Beckmann Rearrangement, Crossed Cannizzaro reaction.	
27	II	Cis-azobenzene, Benzopinacol, Maleic acid, cis-stilbene,	
28	II	Applications: Photochemical cycloaddition reactions, Paterno-Buchi reaction,	
29	II	Photoinduced substitution, Photochlorination,	
30	II	Photochemical reaction in solid state	
31	IV	Transformations using sonication	
32	IV	Benzyl cyanide, biphenyl, cannizzaro reaction,	
33	IV	cinnamaldehyde, cyclohexanone, ethyl phenyl ether,	
34	IV	Applications: Hydrolysis of nitriles, solvolysis,	
35	IV	Strecker synthesis, Reformatsky reaction,	
36	IV	Curtius rearrangement,	
37	IV	oxymercuration of olefins,	
38	IV	Dieckmann cyclisation,	
39	IV	Isomerization of maleic acid to fumaric acid.	
40	IV	Overview on all topics	

Signature of Faculty Member:

Date:

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VEER SURENDRA SAI UNIVERSITY OF TECHNOLOGY, BURLA

Semester: 5th M.Sc

Lesson Plan

Subject: CIM (CH-315)

Session: **Odd 2016-17**

Theory

Course: **5-Year Int.M.Sc (Chemistry)**

Name of the Faculty Member: **Dr A. K. Panda**

Period	Module/ Number	Topics to be Covered	Remarks/Sign. of Faculty Member
1	I	<i>Fuel and combustion:</i> Introduction, classification	
2	I	Calorific value, Petroleum and coal based chemicals	
3	I	Composition of petroleum, fractional distillation	
4	I	Thermal and catalytic cracking of petroleum	
5	I	Fractional distillation of coal tar	
6	I	Fractional distillation of coal tar	
7	I	Diesel engine fuels, Biodiesel	
8	I	Composition, preparation and uses of water gas	
9	I	Composition, preparation and uses of CNG	
10	I	Composition, preparation and uses ofLPG	
11	III	Polymers: Functionality, degree of polymerisation	
12	III	Polymerization mechanisms,	
13	III	Addition or chain growth polymerisation with example	
14	III	Condensation or step growth polymerisation	
15	III	Application of polymers	
16	III	Dyes and pigments: Properties and requirement	
17	III	Types of dyes based on the structure	
18	III	Types of dyes based on application	
19	III	Paints, composition, uses	
20	III	Varnishes, composition, uses	
21	II	Oils and fats: difference,	
22	II	properties of oil	
23	II	Solvent extraction of oils, hydrogenation of oil,	
24	II	Analysis of oil	
25	II	Soap, type, composition,	
26	II	use of oil in the manufacturing of soap	
27	II	Detergents, Definition, classification,	

28	II	Manufacturing of detergents	
29	II	Surface-active agents	
30	IV	Cleansing action of soap and detergents.	
31	IV	Components of cement, classification of cement, raw materials for cement manufacture	
32	IV	Preparation of cement by wet and dry process	
33	IV	Reactions in the kiln	
34	IV	Additives in cements, setting of cement.	
35	IV	Metal extraction I	
36	IV	Metal extraction II	
37	IV	Metal extraction III	
38	IV	Pesticides: DDT manufacture	
39	IV	BHC manufacture	
40	IV	Parathion manufactur	
Signature of Faculty Member:			
ate:		Counter Signature of H.O.D.	

VEER SURENDRA SAI UNIVERSITY OF TECHNOLOGY, BURLA			
Semester: 6 th M.Sc		Lesson Plan	Subject: Natural products (CH-321)
Session: Even 2016-17		Theory	
Course: 5-Year Int.M.Sc (Chemistry)		Name of the Faculty Member: Dr Sukalyan Dash	
Period	Module/ Number	Topics to be Covered	Remarks/Sign. of Faculty Member
1	I	Natural Colouring Matter, General Classification	
2	I	Synthesis of Anthocyanins (Cyanine),	
3	I	Synthesis of Anthocyanins (Cyanine),	
4	I	Synthesis of Flavones (Chryosin) and Flavanol (Querecetin)	
5	I	Synthesis of Flavones (Chryosin) and Flavanol (Querecetin)	
6	I	Porphyrin: Structure, Spectral Properties, Biological Importance	
7	I	Porphyrin: Structure, Spectral Properties, Biological Importance	
8	I	Synthesis of Haemoglobin	
9	I	Synthesis of Chlorophyll-A	
10	I	Synthesis of Chlorophyll-A	

11	II	General Method of Structure Elucidation of Alkaloids	
12	II	General Method of Structure Elucidation of Alkaloids	
13	II	Structure Elucidation of Quinine	
14	II	Structure Elucidation of Quinine	
15	II	Biological Importance and Synthesis of Morphine	
16	II	<i>Vitamins</i> : Introduction	
17	II	Synthesis and Biochemical Importance of Vitamin B ₁ (Thiamine), Vitamin H (Biotin)	
18	II	Synthesis and Biochemical Importance of Vitamin B ₁ (Thiamine), Vitamin H (Biotin)	
19	II	Synthesis and Biochemical Importance of α -Tocopherol (Vitamin E), and Vitamin C	
20	II	Synthesis and Biochemical Importance of α -Tocopherol (Vitamin E), and Vitamin C	
21	III	Introduction and Stereochemistry of Steroids	
22	III	Structure Elucidation of Cholesterol	
23	III	Structure Elucidation of Cholesterol	
24	III	Structure Elucidation of Cholesterol	
25	III	Structure Elucidation of Cholesterol	
26	III	Biological Importance of Bile Acid	
27	III	Biological Importance of Androgens	
28	III	Biological Importance of Oestrogens, Gestrogens	
29	III	Biological Importance of Adrenocortical Hormones	
30	III	Biological Importance of Cortisone	
31	IV	<i>Terpenoids</i> : Classification	
32	IV	Nomenclature	
33	IV	General Methods of Structure Determination	
34	IV	General Methods of Structure Determination	
35	IV	Structure Elucidation of Abietic Acid	
36	IV	Structure Elucidation of Abietic Acid	
37	IV	Structure Elucidation of Abietic Acid	
38	IV	Synthesis and Biological Importance of Farnesol	

39	IV	Synthesis and Biological Importance of Zingerone	
40	IV	Synthesis and Biological Importance of Squalene	
Signature of Faculty Member:			
Date:		Counter Signature of H.O.D.	

VEER SURENDRA SAI UNIVERSITY OF TECHNOLOGY, BURLA

Semester: 6th M.Sc

Lesson Plan

Subject: Basic physical Chemistry-II (CH-324)

Session: **Even 2016-17**

Theory

Course: **5-Year Int.M.Sc (Chemistry)**

Name of the Faculty Member: **Dr Sukalyan Dash**

Period	Module/ Number	Topics to be Covered	Remarks/Sign. of Faculty Member
1	I	Natural Colouring Matter, General Classification	
2	I	Synthesis of Anthocyanins (Cyanine),	
3	I	Synthesis of Anthocyanins (Cyanine),	
4	I	Synthesis of Flavones (Chryosin) and Flavanol (Quercetin)	
5	I	Synthesis of Flavones (Chryosin) and Flavanol (Quercetin)	
6	I	Porphyrin: Structure, Spectral Properties, Biological Importance	
7	I	Porphyrin: Structure, Spectral Properties, Biological Importance	
8	I	Synthesis of Haemoglobin	
9	I	Synthesis of Chlorophyll-A	
10	I	Synthesis of Chlorophyll-A	
11	II	General Method of Structure Elucidation of Alkaloids	
12	II	General Method of Structure Elucidation of Alkaloids	
13	II	Structure Elucidation of Quinine	
14	II	Structure Elucidation of Quinine	
15	II	Biological Importance and Synthesis of Morphine	
16	II	<i>Vitamins</i> : Introduction	
17	II	Synthesis and Biochemical Importance of Vitamin B ₁ (Thiamine), Vitamin H (Biotin)	
18	II	Synthesis and Biochemical Importance of Vitamin B ₁ (Thiamine), Vitamin H (Biotin)	
19	II	Synthesis and Biochemical Importance of α -Tocopherol (Vitamin	

		E), and Vitamin C	
20	II	Synthesis and Biochemical Importance of α -Tocopherol (Vitamin E), and Vitamin C	
21	III	Introduction and Stereochemistry of Steroids	
22	III	Structure Elucidation of Cholesterol	
23	III	Structure Elucidation of Cholesterol	
24	III	Structure Elucidation of Cholesterol	
25	III	Structure Elucidation of Cholesterol	
26	III	Biological Importance of Bile Acid	
27	III	Biological Importance of Androgens	
28	III	Biological Importance of Oestrogens, Gestrogens	
29	III	Biological Importance of Adrenocortical Hormones	
30	III	Biological Importance of Cortisone	
31	IV	<i>Terpenoids</i> : Classification	
32	IV	Nomenclature	
33	IV	General Methods of Structure Determination	
34	IV	General Methods of Structure Determination	
35	IV	Structure Elucidation of Abietic Acid	
36	IV	Structure Elucidation of Abietic Acid	
37	IV	Structure Elucidation of Abietic Acid	
38	IV	Synthesis and Biological Importance of Farnesol	
39	IV	Synthesis and Biological Importance of Zingeberine	
40	IV	Synthesis and Biological Importance of Squalene	

Signature of Faculty Member:

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VEER SURENDRA SAI UNIVERSITY OF TECHNOLOGY, BURLA

Semester: **Sixth**

LESSON PLAN

Subject: **Principles of Inorganic Chemistry (CH-323)**

Session: **Even 2016–2017**

Theory/Sessional

Branch/Course: **Int. M. Sc.**

Name of the Faculty Member: **Dr. Biswa Nath Ghosh**

Period	Module/ Number	Topics to be Covered	Remarks/Sign. of Faculty Member
1-2	I	Theory of Acid-Bases: Arrhenius Theory, Solvent-System Definition, Brönsted-Lowry Theory: Conjugate Acid-Base Pairs	
3-4	I	Lewis Theory, Usanovich Concept, Lux-Flood Concept.	
5-6	I	Acid-base Equilibrium, Strength of Brönsted Acids and Bases: Gas-Phase Proton Affinity	
7-8	I	Acid and Bases in Water, Levelling Effect of Water. Lewis Acid and Bases: Properties	
9-10	I	Hard and Soft Acids and Bases (Pearson's Classification, HSAB Principle), Symbiosis Effect, Strength of Lewis Acids and Bases.	
11-12	I	Super Acids: Hammett-Acidity Function.	
13-14	II	Basic Concepts, Redox Reactions and Electromotive Force, Electrochemical Cells	
15-16	II	Type of Electrodes, Electrode Potential: Standard Electrode Potential, Formal Potential.	
17-18	II	Factors Influencing Electrode Potential: Effect of Concentration, pH, Precipitation, Complex Formation.	
19-20	II	Application of Electrode Potential: Electrochemical Series, Redox Stability in Water	
21-22	II	Redox Potential Diagram: Latimer Diagram (Disproportionation Reaction: Chlorine, Manganese System, Copper System, Oxidation by Atmospheric Oxygen)	
23-24	II	Redox Potential Diagram: Frost Diagram (Manganese), Redox Indicator	
25-26	II	Redox Titration: Titration of Fe(II) by KMnO_4 , Titration of Fe(II) by $\text{K}_2\text{Cr}_2\text{O}_7$.	
27-28	III	Electron Deficient, Electron Precise, and Electron Rich Compounds, Gillespie and Nyholm Rules for Predicting Structures	
29-30	III	Homocyclic Rings of S, Se and Te, S, SN and SeN Compounds- Neutral Sulfur Rings	

31-32	III	Cyclic Sulfur Imides, Sulfur-Nitrogen Rings and Chains (SN _x)	
33-34	III	atenation and Heterocatenation, Heteropoly and Isopoly Anions,	
35-36	III	Electron Counting Rules for Clusters: Wade-Mingos-Lauher Rules and Extended 18 Electron Rules (mno rules), Boranes: Boron Cage Compounds- Closo, Nido, Arachno,	
37-38	III	Heteroboranes, Carboranes, Metalloboranes and Metallocarboranes	
39-40	III	Cage Compounds of S and P, Carbonyl and Carbide Clusters	

Signature of Faculty Member:

Date:

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VEER SURENDRA SAI UNIVERSITY OF TECHNOLOGY, BURLA**LESSON PLAN**Semester: 6th

Session: Even 2016–2017

Branch/Course: Int. MSc

Subject: Biomolecules (CH-325)

Theory/Sessional

Name of the Faculty Member: Dr. Ramakrishna D.S.

Period	Module/ Number	Topics to be Covered	Remarks/Sign. of Faculty Member
1	I	Introduction, Functions of Carbohydrates, Classification of carbohydrate	
2	I	D&L-Isomers, Epimers	
3	I	Structure of Glucose (open chain structure)	
4	I	Cyclic structure of Glucose, Drawbacks of open chain structure	
5	I	Structure of Fructose (both open chain & ring structure)	
6	I	Mutarotation & its Mechanism	
7	I	Reaction of Glucose, Osazone formation	
8	I	Reaction of Fructose, Conversion of Aldose to Ketose & vice versa, Epimerization	
9	I	Chain lengthening & chain shortening of Aldose, Anomers, Chemistry of Disaccharides	
10	I	Sucrose & Lactose (Structure and Properties)	
11	II	Amino acids: (Introduction)	
12	II	structural features,	
13	II	Physical properties of amino acids,	
14	II	optical activity and isoelectric point,	
15	II	essential amino acids,	
16	II	non-essential amino acids	
17	II	Synthesis of alpha amino acids.	
18	II	Chemical properties of alpha amino acids.	

19	II	Chemical properties of alpha amino acids.	
20	II	Overall view on module II	
21	III	Proteins:	
22	III	Peptides	
23	III	Structure determination. Polypeptides or proteins:	
24	III	Classification	
25	III	primary structure of proteins	
26	III	Secondary and tertiary structure of proteins	
27	III	quaternary structure of proteins	
28	III	glycoproteins, denaturation	
29	III	Folding enzymes.	
30	III	Overall view on module III	
31	IV	Introduction , Chemistry of Nucleic Acid , Assignments	
32	IV	Structure of Nucleotides ,Nucleosides ,RNA, DNA	
33	IV	Nitrogenous bases, pentose Sugar, binding of Nucleotide components	
34	IV	Structure of DNA	
35	IV	Structure of RNA ,Types of RNA	
36	IV	m-RNA , t-RNA	
37	IV	Transcription & Translation	
38	IV	Post Transcription Modification , Genetic Code	
39	IV	Central dogma of life , Replication of DNA	
40	IV	Protein Biosynthesis ,genetic error ,Wobblehypothesis	

Signature of Faculty Member:

Date:

Counter Signature of H.O.D.

VEER SURENDRA SAI UNIVERSITY OF TECHNOLOGY, BURLA

LESSON PLAN

Semester: 1st & 7th

Subject: **Group Theory and Wave Mechanics (CH-411)**

Session: Odd 2016–2017

Theory/Sessional

Branch/Course: **M.Sc. (IC) and Int. M.Sc. (Chemistry)**

Name of the Faculty Member: **Dr. Aruna Kumar Barick**

Period	Module/ Number	Topics to be Covered	Remarks/Sign. of Faculty Member
41	I	Symmetry Elements and Symmetry Operations, Matrix Representation of Symmetry Operation, Classes of Operations	
42	I	Point Groups (C_n , C_{nv} , C_{nh} , S_n , D_n , D_{nd} , and D_{nh})	
43	I	Point Groups (T_d , O_h , $D_{\infty d}$, $C_{\infty v}$, and $D_{\infty h}$)	
44	I	Properties of Point Groups,	
45	I	Irreducible and Reducible Representation, Bases of Representation	
46	I	Character of a Representation, Reduction Formula	
47	I	The Great Orthogonality Theorem (Without Proof) and Its Explanation	
48	I	Construction of Character Tables for C_{2v} and C_{3v}	
49	I	Construction of Character Tables for T (Cubic), C_4 (Cyclic) and D_∞ Groups	
50	I	Projection Operator and Direct Product	
51	II	Postulates of Quantum Mechanics, Quantum Mechanical Operators	
52	II	Application of Schrodinger Wave Equation to Particle in a Box	
53	II	Harmonic Oscillator	
54	II	Rigid Rotator	
55	II	Hydrogen Atom, Transformation of Co-ordinates	
56	II	Separations of Variables	
57	II	ϕ , θ and R Equations, Spherical Harmonics	
58	II	Shapes of s , p and d Orbital	
59	II	Probability Density in 1s Orbital, Physical Interpretation of Hydrogen Orbitals	

60	II	Radial Distribution Function and Curves	
61	III	Definition	
62	III	Generalized Angular Momentum	
63	III	Eigen Functions and Eigen Values of Angular Momentum	
64	III	Operator using Ladder Operators	
65	III	Addition of Angular Moments	
66	III	Mutual Interaction of Electron Orbitals and Resultant Vectors	
67	III	Russel–Saunders’s Coupling	
68	III	j–j Coupling	
69	III	Ground State Term Symbols and Hund’s Rule	
70	III	Micro States and Derivation of Russel–Saunders’s Term for P ² , d ² and pd Configuration	
71	IV	Variation Theorem and its Application to Hydrogen atom in Derivation of its Ground State Energy	
72	IV	Perturbation Theory (First Order and Non-degenerate)	
73	IV	Secular Equations	
74	IV	Linear Combination of Atomic Orbitals (LCAO) Approximation (Molecular Orbital Theory)	
75	IV	Application to Hydrogen Molecule Ion	
76	IV	Huckel Theory of Conjugated Systems	
77	IV	Bond Order and Charge Density Calculations	
78	IV	Applications to Ethylene, Butadiene	
79	IV	Applications to Cyclopropenyl Radical, Cyclobutadiene, etc.	
80	IV	Spin and Anti-symmetric Nature of Wave Function (Pauli’s Exclusion Principle)	

Signature of Faculty Member:

Date:

Counter Signature of H.O.D.

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VEER SURENDRA SAI UNIVERSITY OF TECHNOLOGY, BURLA

LESSON PLAN

Semester: 1st & 7th

Subject: Structure and Reactivity (CH-413)

Session: Odd 2016–2017

Theory/Sessional

Branch/Course: M.Sc. (IC) and Int. M.Sc. (Chemistry)

Name of the Faculty Member: Dr. Sukalyan Dash

Period	Module/ Number	Topics to be Covered	Remarks/Sign. of Faculty Member
1	I	Delocalized Chemical Bonding, Conjugation	
2	I	Cross Conjugation	
3	I	Resonance	
4	I	Hyperconjugation, Bonding in Fullerenes, Tautomerism.	
5	I	Aromaticity in Benzenoid and Non-benzenoid Compounds, Alternant and Non-alternant Hydrocarbons	
6	I	Huckel's Rule, Energy Levels of π -molecular Orbitals of Simple Systems, Annulenes, Anti-aromaticity, Homo-aromaticity	
7	I	Bonds Weaker than Covalent (Addition compounds)	
8	I	Crown Ether Complexes and Cryptands	
9	I	Inclusion Compounds, Cyclodextrins	
10	I	Catenanes and Rotaxanes	
11	II	Types of Mechanisms, Types of Reactions	
12	II	Thermodynamic and Kinetic Requirements	
13	II	Kinetic and Thermodynamic Control	
14	II	Hammond's Postulate, Curtin-Hammett Principle	
15	II	Potential Energy Diagrams, Transition States and Intermediates	
16	II	Methods of Determining Mechanisms	
17	II	Methods of Determining Mechanisms	
18	II	Hard and Soft Acids and Bases	

19	II	Hammett Equation and Linear Free Energy Relationship	
20	II	Substituent and Reaction Constants. Taft Equation	
21	III	Non-classical Carbocations	
22	III	Generation and Structure of Free Radicals	
23	III	Generation and Structure of Carbenes	
24	III	Generation and Structure of Nitrenes	
25	III	Generation and Structure of Arynes	
26	III	General Discussion on Isotope Effect	
27	III	General Discussion on Isotope Effect	
28	III	Stereoselective, Regioselective Reactions	
29	III	Stereospecific and Regiospecific Reactions	
30	III	Stereospecific and Regiospecific Reactions	
31	IV	S_N^2 , S_N^1 Mechanisms	
32	IV	Mixed S_N^1 and S_N^2 Mechanism	
33	IV	SET Mechanisms. The Neighboring Group Mechanism	
34	IV	Neighboring Group Participations by Sigma and Pi Bonds	
35	IV	Classical and Non-classical Carbocations, Phenonium Ions, Norbornyl System	
36	IV	Nucleophilic Substitution at Allylic, Carbon	
37	IV	Nucleophilic Substitution at Aliphatic Trigonal and Vinylic Carbon	
38	IV	Effects of Substrate Structure, Attacking Nucleophile	
39	IV	Effects of Leaving Group and Reaction Medium	
40	IV	Phase Transfer Catalysis, Ambident Nucleophile, Regioselectivity	

Signature of Faculty Member:

Date:

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VEER SURENDRA SAI UNIVERSITY OF TECHNOLOGY, BURLA

Semester -7th & 1st M.Sc
M.Sc

LESSON PLAN

Subject: Thermodynamics & Chemical dynamics (CH-414)

Session: 2016

Theory

Branch/Course: Integrated M.Sc & M.Sc

Name of the Faculty Member: **Dr Trinath Biswal**

Period	Module /Number	Topic to be covered
1	I	Laws of Thermodynamics
2	I	Free Energy
3	I	Partial Molar Properties & Chemical Potential
4	I	Third Law of Thermodynamics & Determination of Entropy
5	I	Entropy & Probability
6	I	Boltzmann-Planck equation
7	I	Partial free energy ,molar volume ,molar heat content
8	I	Problems
9	I	Fugacity & its Determination
10	I	Determination of fugacity by Graphical Method
11	I	Determination of fugacity by Approximate & General Method
12	II	Probability Distribution, Ensemble averaging & its types
13	II	Postulates of Ensemble averaging , Canonical , Grand Canonical & Microcanonical
14	II	Corresponding Distribution Law
15	II	Translational & Rotational Partition Function
16	II	Vibrational Partition Function, Electronic Partition Function
17	II	Calculation of thermodynamic properties of Partition Function
18	II	Applications of Partition Function
19	II	Behavior of solids , Fermi- Dirac Statistics
20	II	Chemical equilibria & equilibrium constant in terms of Partition function
21	II	Bose-Einstein Statistics , Distribution Law & application to He
22	II	Phosphate group transfer & ATP , Biological oxidation & reduction Reaction
23	III	Interionic attraction Theory & Debye- Huckel Treatment
24	III	Onsager Limiting law its Verification & Modification
25	III	Activity & Activity coefficient
26	III	Debye- Huckel – Bronsted Equation , Salt effect
27	III	Primary salt effect & secondary salt effect
28	III	Determination of activity coefficient by solubility Method , Ion Association
29	III	Determination of thermodynamic dissociation constant of weak electrolytes by Shedlovsky Method & EMF Method

30	III	Amino acids ,Hydrogen ion concentration	
31	III	Ampholytes Isoelectronic points	
32	IV	Introduction ,collision theory of Reaction Rate	
33	IV	Theory of Absolute Reaction rate of both unimolecular & Bimolecular Reaction	
34	IV	Lindeman Mechanism	
35	IV	Arrehenious theory & Activated complex	
36	IV	Reaction between ions	
37	IV	Stedy-State Kinetics &its Problems , Examples	
38	IV	Dynamic chain reactions of Hydrogen & Bromine	
39	IV	Pyrolysis of Acetaldehyde ,Ethane	
40	IV	Fast reaction &its study by Relaxation Method	
41	IV	Flash photolysis & NMR Technique	

Signature of Faculty member:

Date:

Counter Signature of H.O.D

VEER SURENDRA SAI UNIVERSITY OF TECHNOLOGY, BURLA			
Semester - 1 st & 7 th		LESSON PLAN	Subject: <u>Polymer Chemistry (CH-415)</u>
Session: 2016		Theory/ Sessional	
Branch/Course: M.Sc & Integrated M.Sc		Name of the Faculty Member: Dr Trinath Biswal	
Period	Module /Number	Topic to be covered	
1	I	Introduction, classification of polymer, DP, Tacticity,	
2	I	Functionality, crystallinity , Degree of crystallinity	
3	I	Glass Transition temperature &its application	
4	I	Inorganic polymer & Elemento -Organic polymer ,Crystallisability	
5	I	Chain-growth polymerization ,its mechanism , Kinetics of Chain –growth polymerization	
6	I	Polyaddition polymerization , Step-growth polymerization ,its mechanism , Kinetics of step – growth polymerization	
7	I	Distinction between Step-growth & Chain-growth polymerization , Molecular weight control &Molecular weight distribution of linear polymer	
8	I	Polyfunctional step polymerization, Newer type of step polymerization	
9	I	Radical chain polymerization ,Molecular weight , chain transfer, Inhabitation &Retardation	
10	I	Determination of absolute rate constant, Auto acceleration	
11	II	Emulsion Polymerization ,Qualitative aspect , Mechanism	
12	II	CMC , Salient features, Application, Advantages	
13	II	Ionic chain polymerization ,comparison of radical & ionic chain polymerization	
14	II	Kinetics of cationic chain polymerization ,Carother equation	
15	II	Kinetics of anionic chain polymerization ,features of cationic &anionic chain polymerization	

16	II	Cationic polymerization & Anionic polymerization of carbon-carbon double bond	
17	II	Block copolymer ,types synthesis ,properties & Application	
18	III	Chain copolymerization ,classification properties , application	
19	III	Radical chain co-polymerization &its kinetic study	
20	III	Ionic chain co-polymerization &its kinetic study	
21	III	Ring opening polymerization , Examples & Mechanism	
22	III	General characteristic Ring opening polymerization of cyclic ethers & cyclic amides	
23	III	Stereoisomerism ,types of stereoisomerism in polymers	
24	III	.Properties of Stereo- regular polymers	
25	III	Forces of Stereo -regulation in alkene polymerization	
26	III	Ziegler-Natta polymerization ,Types ,Mechanism & kinetic study	
27	III	Crystalline melting point, Glass transition temperature	
28	III	.Relationship of T_g with molecular weight , plasticizer ,n copolymer	
29	III	Properties involving Large deformation	
30	IV	Properties involving small deformation	
31	IV	Ideal ,Azeotropic , Alternating Copolymerization	
32	IV	Property requirement & Utilization of polymers	
33	IV	Fracture & Deformation of polymer	
34	IV	,Fracture Mechanism , Types of Fracture	
35	IV	Crack growth & its Mechanism	
36	IV	Cyclic deformation &its Molecular aspects	
37	IV	Healing of polymer. Adhesives	
38	IV	Conducting polymer , characteristics & Examples	
39	IV	Different types of Molecular weight	
40	IV	Molecular weight Determination	
41	IV	Molecular weight Determination & Short questions	

Signature of Faculty member:

Date:

Counter Signature of H.O.D

VEER SURENDRA SAI UNIVERSITY OF TECHNOLOGY, BURLASemester: **2nd, 8th****LESSON PLAN**Subject: **Stereochemistry (CH-422)**Session: **Even 2016–2017****Theory/Sessional**Branch/Course: **M.Sc and**Name of the Faculty Member: **Dr. Ramakrishna D.S.****Int. MSc**

Period	Module/ Number	Topics to be Covered	Remarks/Sign. of Faculty Member
1	I	Chirality, Fischer projection and R and S notations,	
2	I	Threo and erythro nomenclature, E and Z nomenclature,	
3	I	Optical isomerism in biphenyls	
4	I	Allenes,	
5	I	Concept of Prostereoisomerism	
6	I	Assymmetric synthesis	
7	I	Assymmetric synthesis (enzymatic)	
8	I	catalytic nexus	
9	I	Conformation of a few acyclic molecules (5memb)	
10	I	Conformation of a few acyclic molecules (6memb)	
11	II	Conformation of a few acyclic molecules (hetero)	
12	II	Conformation of cyclic systems having one sp ² carbon atoms	
13	II	two sp ² carbon atoms	
14	II	Dynamic stereochemistry	
15	II	Conformation and reactivity, Selection of substrates	
16	II	Quantitative correlation between conformation and reactivity	
17	II	Weinstein-Eliel equations and Curtin-Hammett principles	
18	II	Conformational effects on stability and reactivity in acyclic compounds	
19	II	Ionic elimination	
20	II	Intramolecular rearrangements	
21	II	Neighbouring group participation	

22	II	cyclic systems	
23	II	Nucleophilic substitution reaction at ring carbon	
24	II	Formation and Cleavage of epoxide rings	
25	II	Addition reactions to double bonds	
26	II	Elimination reactions	
27	III	Molecular dissymmetry	
28	III	chiroptical properties	
29	III	linearly and circularly polarised lights	
30	III	circular birefringence and circular dichroism	
31	IV	ORD, Plane curves, Cotton effect	
32	IV	Rotatory Dispersion of ketones	
33	IV	the Axial Haloketone rule	
34	IV	the Octane rule	
35	IV	Helicity rule	
36	IV	Lowe's rule	
37	IV	Empirical rule	
38	IV	Benzene chromophore.	
39	IV	Problems	
40	IV	Overall view	

Signature of Faculty:

Date:

Counter Signature of H.O.D.

VEER SURENDRA SAI UNIVERSITY OF TECHNOLOGY, BURLA

LESSON PLAN

Semester: 2nd & 8th

Subject: Spectroscopy-I (CH-423)

Session: Even 2016–2017

Theory/Sessional

Branch/Course: M.Sc. (IC) and Int. M.Sc. (Chemistry)

Name of the Faculty Member: **Dr. Aruna Kumar Barick**

Period	Module/ Number	Topics to be Covered	Remarks/Sign. of Faculty Member
1	I	Atomic Spectroscopy: Introduction	
2	I	Electromagnetic Spectrum	
3	I	General Discussion on Various Molecular Excitation Processes	
4	I	Spectra of Hydrogen and Hydrogen Like Atoms	
5	I	Alkali Metals Spectra	

6	I	L-S Coupling, Term Symbols,	
7	I	Space Quantization	
8	I	Zeeman Effect and Stark Effect	
9	I	Paschen-Bach Effect	
10	I	Problem Solve	
11	II	<i>Rotational and Vibrational Spectroscopy:</i> Introduction	
12	II	Rotation of Molecules and Classification of Molecules	
13	II	Molecular Spectra of Diatomic Gases	
14	II	Rotational Spectra of Diatomic Molecules	
15	II	Rotational Spectra of Polyatomic Molecules	
16	II	Vibrational Spectra of Diatomic Molecules	
17	II	Intensity of Spectral Lines	
18	II	Vibrational-Rotational Spectra	
19	II	P, Q, and R Branches	
20	II	Problem Solve	
21	III	<i>Raman Spectroscopy:</i> Introduction	
22	III	Quantum Theory of Raman Effect	
23	III	Classical Theory of Raman Effect	
24	III	Pure Rotational Raman Spectra	
25	III	Vibrational Raman Spectra	
26	III	Polarization of Light and the Raman Effect	
27	III	Rotational-Vibrational Raman Spectra	
28	III	Comparison with IR spectra	
29	III	Structure Determination from Raman and IR Spectra	
30	III	Problem Solve	
31	IV	<i>Photoelectron Spectroscopy:</i> Basic Principles	
32	IV	Photoelectric effect, Ionization Process, Koopman's	

		Theorem	
33	IV	Photoelectron Spectra of Simple Molecules	
34	IV	ESCA, Chemical Information from ESCA	
35	IV	Auger Electron Spectroscopy – Basic Idea	
36	IV	<i>Mössbauer Spectroscopy</i> : Introduction	
37	IV	Principles of Mossbauer Spectroscopy	
38	IV	Experimental Methods, Theoretical Aspects	
39	IV	Quadrupole Splitting, Magnetic Hyperfine Interaction	
40	IV	Problem Solve	
<p>Signature of Faculty Member:</p> <p>Date:</p> <p style="text-align: right;">Counter Signature of H.O.D.</p>			

<p>VEER SURENDRA SAI UNIVERSITY OF TECHNOLOGY, BURLA</p> <p>LESSON PLAN</p>			
Semester: 2 nd & 8 th		Subject: Organic Reaction Mechanism (CH-424)	
Session: Even 2016–2017		Theory/Sessional	
Branch/Course: M.Sc. (IC) and Int. M.Sc. (Chemistry)		Name of the Faculty Member: Dr. Sukalyan Dash	
Period	Module/ Number	Topics to be Covered	Remarks/Sign. of Faculty Member
1	I	S _E ¹ , S _E ² and S _E ⁱ Mechanisms	
2	I	Effect of Substrate, Leaving Group and Solvent	
3	I	Reactions Hydrogen Exchange, Migration of Double Bonds, Keto-Enol Tautomerism	
4	I	Halogenation, Aliphatic Diazonium Coupling, Stork-Enamine Reaction	
5	I	Structure Reactivity, Relationship in Mono-substituted Benzene	
6	I	Orientation in Benzene Ring with More than One Substituent, Vilsmeier-Haack Reaction, Pechmann Reaction	
7	I	Introduction, Mechanisms of Aromatic Nucleophilic Substitutions (S _N Ar, S _N ¹ , Aryne)	

8	I	Effect of Substrates, Leaving Groups, and Nucleophile	
9	I	Reactions: Nucleophilic Displacement in Arenodiazonium Salts by Different Nucleophiles, Chichibabin Reaction	
10	I	Reactions: Nucleophilic Displacement in Arenodiazonium Salts by Different Nucleophiles, Chichibabin Reaction	
11	II	Electrophilic Addition	
12	II	Nucleophilic Addition	
13	II	Free Radical Addition	
14	II	Orientation and Reactivity, Addition to Cyclopropanes	
15	II	Hydroboration, Michael Reaction, Sharpless Asymmetric Epoxidation	
16	II	Mechanism and Reactivity	
17	II	Mannich Reaction, LiAlH ₄ Reduction of Carbonyl Compounds, Acids, Esters, Nitriles	
18	II	Addition of Grignard Reagents, Reformatsky Reaction, Aldol Condensation	
19	II	Knoevenagel Condensation, Perkin Reaction, Tollens Reaction	
20	II	Wittig Reaction, Prins Reaction, Benzoin Condensation	
21	III	E ² Mechanism	
22	III	E ¹ Mechanism	
23	III	E ¹ _{CB} Mechanism	
24	III	Orientation, Effect of Substrate	
25	III	Effect of Base, Leaving Group	
26	III	Effect of Medium, Orientation of Double Bond	
27	III	Saytzeff and Hoffman Rules, Pyrolytic Elimination Reaction	
28	III	Oxidative Elimination (Oxidation of Alcohol by Chromium, Moffatt Oxidation)	
29	III	Cleavage of Quaternary Ammonium Hydroxides	

30	III	Chugaev Reaction, Shapiro Reaction	
31	IV	Nature of Migration	
32	IV	Migratory Aptitude	
33	IV	Memory Effects	
34	IV	Wagner-Meerwein Rearrangement	
35	IV	Favorskii Rearrangement	
36	IV	Arndt-Eistert Synthesis	
37	IV	Neber Rearrangement	
38	IV	Hofmann Rearrangement	
39	IV	Baeyer-Villiger Rearrangement	
40	IV	Sommelet-Hauser Rearrangement	

Signature of Faculty Member:

Date:

Counter Signature of H.O.D.

VEER SURENDRA SAI UNIVERSITY OF TECHNOLOGY, BURLA

LESSON PLAN

Semester: 2nd and 8th

Subject: SCNC

Session: Even 2016–2017

Theory/Sessional

Branch/Course: M.Sc. (IC) and Int. M.Sc. (Chemistry)

Name of the Faculty Member: Dr. Monalisa Mohapatra

Period	Module/Number	Topics to be Covered	Remarks/Sign. of Faculty Member
1	I	Derivation of phase rule, Brief concept on one and two component system	
2	I	Water, Sulphur system	
3	I	Application of phase rule to three component systems of solids	
4	I	Application of phase rule to three component systems of liquids	
5	I	Kinetics of Catalytic Reactions	
6	I	Acid-base Catalysis, Enzyme Catalysis	
7	I	Heterogeneous Catalysis	
8	II	Surface tension	
9	II	Capillary action	

10	II	Adsorption, types of adsorption	
11	II	Gibbs adsorption isotherm	
12	II	Freundlich's adsorption isotherm	
13	II	Langmuir's adsorption isotherm and its limitations	
14	II	BET adsorption isotherm and its applications	
15	II	Heat of adsorption	
16	II	Estimation of surface areas of solids from solution adsorption studies	
17	II	Brief concepts on micelle	
18	II	Reversed micelle	
19	II	Microemulsions	
20	III	Polymer-definition, Types of polymer	
21	III	Number average and weight average macromolecules	
22	III	Determination of molecular weights of macromolecules	
23	III	Osmometry	
24	III	Viscometry	
25	III	Diffusion and Light scattering method	
26	III	Kinetics of polymerization	
27	III	Donnan Effect	
28	III	Stereochemistry of polymerization	
29	IV	Classification of nuclides, nuclear stability, binding energy	
30	IV	Nuclear models, Characteristics of radioactive decay	
31	IV	Decay kinetics, parent-daughter decay growth relationships	
32	IV	Detection and measurement of radioactivity, advances in the solid and liquid scintillation counting techniques	
33	IV	Methods for the determination of half life period of single and mixed radionuclides	
34	IV	Nuclear fission, nuclear fuels and nuclear reactors, nuclear fuel reprocessing	
35	IV	Fast breeder reactors, radiological safety aspects and radioactive waste managements	
36	IV	Interaction of radiation with matter, effect of ionizing/ non-ionizing radiations on water	
37	IV	Aqueous solutions and on organic compounds, radiation dosimetry	
38	IV	Preparation and separation of radioactive isotopes	
39	IV	Application of radioisotopes and radiations in various fields	
40	IV	Isotopic dilution techniques, neutron activation analysis and its applications	

Signature of Faculty Member:	
Date:	Counter Signature of H.O.D.

VEER SURENDRA SAI UNIVERSITY OF TECHNOLOGY, BURLA			
		LESSON PLAN	
Semester: 3rd and 9th		Subject: ENVIRONMENTAL Chemistry	
Session: Odd 2016–2017		Theory/Sessional	
Branch/Course: M. Sc. And		Name of the Faculty Member: Prof. R. B. Panda	
Int. M.SC.			
Period	Module/ Number	Topics to be Covered	Remarks/Sign. of Faculty Member
1-2	I	Air Pollutants, Air Quality Standards, Production, Fate, Effects and Control of Gaseous Pollutants	
3-4	I	Oxides of Carbon, Nitrogen and Sulphur, Organic Air Pollutants, Photochemical Reaction	
5-6	I	Photochemical Smog, Greenhouse Effect, Climate Change, Global warming, Acid Rain and Ozone Depletion.	
7-8	I	Water World, Source of Water, Water Quality, Water Pollutants (Inorganic and Organic), Sources, Fate, Effects and Controlling Measures, Chemical Speciation, Pollution by Radionuclides	
9-10	I	Biochemical Oxygen Demand, Chemical Oxygen Demand, Eutrophication, Biodegradation of Pollutants	
11-12	II	Treatment of Water for Drinking, Electro-dialysis, Ion Exchange, Reverse Osmosis	
13-14	II	Desalination Processes, Removal of Iron, Manganese, Phosphorous, Calcium and Nitrogen and Treatment of Water for Industrial Purposes	
15-16	II	Sedimentation, Coagulation, Flocculation, Filtration, Adsorption, Disinfection of Water	
17-18	II	Sewage Treatment (Physical and Chemical Methods), Health Effects of Drinking Water Treatment Technologies	
19-20	II	Impact of Detergents, Pesticides and Other Additives	

		on Sewage Treatment	
21-22	III	Sources of Oil Pollution, Chemistry and Fate of Hydrocarbons Oil in Run Off and Ground Water	
23-24	III	Biodegradation, Effect on Aquatic Organisms and Communities	
25-26	III	Treatment and Disposal Technology	
27-28	III	Soil Pollutants (Inorganic, Organic, Pesticides, Radionuclides)	
29-30	III	Sources and Effects on Nature and Properties of Soil, Crops, Plants and Terrestrial Animals.	
31-32	IV	Nature and Sources of Hazardous Wastes, Classification, Characteristics and Constituents,	
33-34	IV	Transport and Effects, Treatment by Physical and Chemical Methods, Thermal Treatment Methods	
35-36	IV	Biodegradation of Wastes, Disposal of Hazardous Wastes. Waste Management and Industrial by Products, Natural Hazards and Management	
37-38	IV	Control of Subsurface Migration of Hazardous Waste, Biomedical Waste Management	
39-40	IV	Environmental Management and Sustainable Development	
<p>Signature of Faculty Member:</p> <p>Date: _____ Counter Signature of H.O.D. _____</p>			

VEER SURENDRA SAI UNIVERSITY OF TECHNOLOGY, BURLA

LESSON PLAN

Semester: **4th and 10th**

Session: **Even 2016–2017**

Branch/Course: **M.Sc. (IC) and Int MSc**

Subject: **Chemistry of Materials
Theory/Sessional**

Name of the Faculty Member: **Dr. Sukalyan Dash**

Period	Module/ Number	Topics to be Covered	Remarks/Sign. of Faculty Member
1	I	Glassy State, Glass Formers	
2	I	Glass Modifiers, Applications	

3	I	Ceramic Structures	
4	I	Mechanical Properties	
5	I	Clay Products. Refractories	
6	I	Characterizations, Properties and Applications	
7	I	Macroscopic Composites, Dispersion-Strengthened and Particle-Reinforced, Fibre-Reinforced Composites	
8	I	Nanocrystalline Phase, Preparation	
9	I	Procedures, Special Properties	
10	I	Applications	
11	II	Types of Ionic Conductors	
12	II	Mechanism of Ionic Conduction	
13	II		
14	II	Interstitial Jumps (Frenkel), Vacancy Mechanism	
15	II	Diffusion Superionic Conductors	
16	II	Phase Transition and Mechanism of Conduction in Superionic Conductors	
17	II	Phase Transition and Mechanism of Conduction in Superionic Conductors	
18	II	Phase Transition and Mechanism of Conduction in Superionic Conductors	
19	II	Examples and Applications of Ionic Conductors	
20	II	Examples and Applications of Ionic Conductors	
21	III	Conducting Organics, Organic Superconductors	
22	III	Magnetism in Organic Materials	
23	III	Fullerenes-Doped, Fullerenes as Superconductors	
24	III	Molecular Rectifiers	
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26	III	Artificial Photosynthetic Devices	

27	III	Optical Storage Memory and Switches-Sensors	
28	III	Nonlinear Optical Effects, Second and Third Order, Molecular Hyperpolarisability	
29	III	Second Order Electric Susceptibility	
30	III	Materials for Second and Third Harmonic Generation	
31	IV	Introduction to <i>Thin Films and Langmuir-Blodgett Films</i>	
32	IV	Preparation Techniques: Evaporation/Sputtering	
33	IV	Chemical Processes, Sol-Gel Method	
34	IV	Growth Techniques of Langmuir-Blodgett (LB) Film	
35	IV	Photolithography	
36	IV	Properties and Application of Thin and LB Films	
37	IV	Molecular Shape, Structure and Configuration of Polymeric materials	
38	IV	Crystallinity, Stress-Strain Behaviour	
39	IV	Polymer Types and Their Applications	
40	IV	Conducting and Ferroelectric Polymers	

Signature of Faculty Member:

Date:

Counter Signature of H.O.D.